Periodic Elements Need to Knows Worksheet

<u>Directions:</u> Use your periodic table to find the Symbol, atomic number & atomic mass (round the atomic mass to the nearest whole number). Determine if the element is a metal, nonmetal or metalloid and if it is a solid, liquid or a gas at room temperature.

- 1. Knowing that the atomic number tells us the number of protons, fill in the "# protons" column.
- 2. Since the # of electrons is equal to the number of protons in a neutral atom, now fill in the "# electrons" column.
- 3. Knowing the atomic mass = the # of protons + the # of neutrons, simply subtract the # of protons from the atomic mass to determine the number of Neutrons. Now fill in the "# neutron" column with this number you have just calculated.

Element	Symbol	Atomic	Atomic	#	#	#	Solid	Metal
		Number	Mass	- Protons	= Neutrons	Electrons	Liquid Gas	Metaloid Nonmetal
Hydrogen								
Oxygen								
Carbon								
Nitrogen								
Calcium								
Phosphorus								
Sulfur								
Helium								
Sodium								
Chlorine								
Aluminum								
Potassium								







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Element	Symbol	Atomic Number	Atomic Mass	# - Protons	# = Neutrons	# Electrons	Solid Liquid Gas	Metal Metaloid Nonmetal
Hydrogen	Н	1	1	1	0	1	G	NM
Oxygen	O	8	16	8	8	8	G	NM
Carbon	С	6	12	6	6	6	S	NM
Nitrogen	N	7	14	7	7	7	G	NM
Calcium	Ca	20	40	20	20	20	S	M
Phosphorus	P	15	31	15	16	15	S	NM
Sulfur	S	16	32	16	16	16	S	NM
Helium	Не	2	4	2	2	2	G	NM
Sodium	Na	11	23	11	12	11	S	M
Chlorine	Cl	17	35	17	18	17	G	NM
Aluminum	Al	13	27	13	14	13	S	M
Potassium	K	19	39	19	20	19	S	M





